## Smt. A.A.A. Govt. P. G. College, Kalka Department Of Chemistry

Name of Assistant Professor: Dr. Indu & Mrs. Puneet

Class- B. Sc. III (Sem-V)

B.Sc - III	August	Carbohydrates: Classification and nomenclature of Monosaccharides, mechanism
(Organic Chemistry)		of osazone formation, interconversion of glucose and fructose, chain lengthening and chain shortening of aldoses. Configuration of monosaccharides. Erythro and threo diastereomers. Conversion of glucose into mannose. Formation of glycosides, Determination of ring size of glucose and fructose. Open chain and cyclic structure
		of D (+)-glucose & amp; D(-) fructose.
	September	Mechanism of mutarotation. Structures of ribose and deoxyribose. An introduction to disaccharides (maltose, sucrose and lactose) and polysaccharides (starch and cellulose) without involving structure determination.
	October	Organometallic Compounds: Organomagnesium compounds: the Grignard reagents-formation, structure and chemical reactions. Organozinc compounds: formation and chemical reactions. Organolithium compounds: formation and chemical reactions. NMR Spectroscopy: Principle of nuclear magnetic resonance, the PMR spectrum number of signals, peak areas
	November	<b>NMR Spectroscopy:</b> equivalent and nonequivalent protons positions of signals and chemical shift, shielding and deshielding of protons, proton counting, splitting of signals and coupling constants, magnetic equivalence of protons. Discussion of PMR spectra of the molecules: ethyl bromide, n-propyl bromide, isopropyl bromide, 1,1-dibromoethane, ethanol, acetaldehyde, ethyl acetate, toluene, benzaldehyde and acetophenone. Simple problems on PMR spectroscopy for structure determination of organic compounds.
B.Sc - III (Inorganic Chemistry)	August	<b>Metal-Ligand Bonding in Transition Metal complexes:</b> Limitations of valence bond theory, An elementary idea of crystal field theory, crystal field splitting in octahedral, tetrahedral and square planer complexes, factors affecting the crystal field parameters.
	September	<b>Thermodynamics and Kinetic Aspects of metal complexes</b> : A brief outline of thermodynamic stability of metal complexes and factors affecting the stability, Irving William Series, substitution reactions of square planer complexes of Pt[II], Trans effect.
	October	Magnetic properties of Transition metal complexesTypes of magnetic materials, magnetic susceptibility, method of determiningmagnetic susceptibility, spin only formula, L-S coupling, correlation of µs and µeffvalues, orbital contribution to magnetic moments, application of magnetic momentdata for 3d metal complexes.
	November	<b>Electronic spectra of Transition metal complexes</b> Selection rules for d-d transition, spectroscopic ground states, spectrochemical series, orgel energy level diagram for d1 and d9 states, discussion of electronic spectrum of [Ti(H2O)6]+3 complex ion.
B.Sc - III (Physical Chemistry)	October	Quantum Mechanics-I Black-body radiation, Plank's radiation law, photoelectric effect, postulates of quantum mechanics, quantum mechanical operator s, commutation relations, Hamiltonian operator, Hermitian operator, average value of square of Hermitian as a positive quantity, Role of operators in quantum mechanics, To show quantum mechanically that position and momentum cannot be predicated simultaneously, Determination of wave function & energy of a particle in one dimensional box.

		Physical Properties and Molecular Structure Optical activity, polarization – (Clausius – Mossotti equation derivation excluded). Orientation of dipoles in an electric field, dipole moment, induced dipole moment, measurement of dipole moment -temperature method and refractivity method, dipole moment and structure of molecules, Magnetic permeability, magnetic susceptibility and its de termination. Application of magnetic susceptibility, magnetic properties – paramagnetism, diamagnetism and ferromagnetism.
	November	SpectroscopyElectromagnetic radiation, regions of spectrum, basic features of spectroscopy, statement of Born oppenheimer approximation, Degrees of freedom.Rotational SpectrumSelection rules, Energy levels of rigid rotator (semi-classical principles), rotational spectra of diatomic molecules , spectral intensity distribution using population distribution (Maxwell-Boltzmann distribution), determination of bond length and isotopic effect .
		<ul> <li>Vibrational spectrum</li> <li>Selection rules, Energy levels of simple harmonic oscillator, pure vibrational spectrum of diatomic molecules, determination of force constant and qualitative relation of force constant and bond energy, idea of vibrational frequencies of different functional groups.</li> <li>Raman Spectrum</li> <li>Concept of polarizibility, pure rotational and pure vibrational Raman spectra of diatomic molecules, selection rules, Quantum theory of Raman spectra.</li> </ul>